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Accordingly, the present claims are not obvious over Chasalow. Applicants respectfully request that the rejection be withdrawn.

(ii) Claims 12, 13, 16-17, and 20 have been rejected under 35 U.S.C. § 103(a) as obvious over Ansell. The Examiner acknowledges that Ansell does not teach the instant drugs, but concludes that in view of the suggestion that the method is applicable to any active agent, it would have been obvious to one of ordinary skill in the art to use any active agent with a reasonable chance of success.

This rejection is respectfully traversed.

Ansell teaches therapeutic agents capable of being formulated in liposomes or micelles, whereby the therapeutic agent is attached to a fatty acid chain of a phospholipid. As can be seen from col. 6, lines 3-5 of Ansell, the RC(O) group present in the compound of formula (III), and hence also in the compounds of Formula (IV) - (XI) of Ansell, is a fatty acid radical, such as lauroyl, myristoyl, palmitoyl, stearoyl, or oleoyl. In contrast, the compounds of the present invention do not rely upon emulsifiers or liposome or micelle formation in order to increase the solubility and bioavailability of the therapeutic agent. Rather, in the present invention, the therapeutic agent is conjugated to phosphocholine via a linker, inserted between the phosphocholine and an alcohol group on the therapeutic agent. This formulation facilitates enzymatic cleavage of the phosphocholine linker bond and liberates the linker, which then spontaneously eliminates to liberate the therapeutic agent and an inert molecule arising from the decomposed linker.

Based upon the teachings of Ansell, one of ordinary skill in the art would have had no expectation of success that non liposomal or micellar formulated phosphocholine conjugated therapeutic agents could be prepared wherein the therapeutic agent was not conjugated to a fatty acid radical of the phosphocholine.

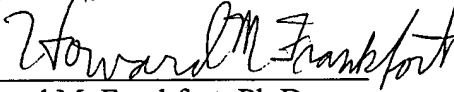
Ansell simply does not teach or suggest the presently claimed conjugates containing the linkers of the present invention attached between the therapeutic agent and the phosphocholine.

Accordingly, the present claims are not obvious over Ansell. Applicants respectfully request that the rejection be withdrawn.

In view of the above amendments and arguments, the pending claims in this application are believed to be in condition for allowance. Accordingly, the Examiner is respectfully requested to pass this application to issue.

Dated: June 25, 2004

Respectfully submitted,

By 

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Registration No.: 32,613

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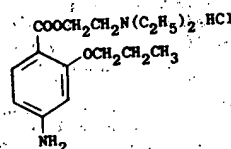
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USE: Insecticide.

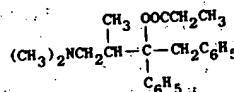
7850. Propoxycaine Hydrochloride. 4-Amino-2-propoxybenzoic acid 2-diethylaminoethyl ester hydrochloride; 2-diethylaminoethyl 4-amino-2-propoxybenzoate hydrochloride; 2-diethylaminoethyl 2-propoxy-4-aminobenzoate hydrochloride; Ravocaine hydrochloride; Pravocaine hydrochloride; Blockaine hydrochloride. $C_{16}H_{21}ClN_2O_5$, mol wt 330.86. C 58.08%, H 8.23%, Cl 10.72%, N 8.47%, O 14.51%. Prep'n: Clinton, Laskowski, U.S. pat. 2,689,248 (1954 to Sterling Drug).



White, odorless crystals, mp 148-150°. Discolors upon prolonged exposure to light and to air. Freely sol in water; sol in ethanol, chloroform. Sparingly sol in ether. Practically insol in acetone, chloroform. pH of a 2% aq soln 5.4.

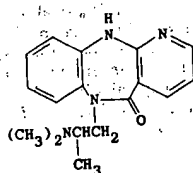
THERAP CAT: Local anesthetic.

7851. Propoxyphene. $[S-(R^*,S^*)]-\alpha$ -[2-(Dimethylamino-1-methylethyl)- α -phenylbenzeneethanol propanoate (ester); α -d-4-dimethylamino-3-methyl-1,2-diphenyl-2-butanol propionate; (+)-1,2-diphenyl-2-propionyloxy-3-methyl-4-dimethylaminobutane; (+)-4-dimethylamino-1,2-diphenyl-3-methyl-2-propionyloxybutane; d-propoxyphene; dextro-propoxyphene. $C_{27}H_{35}NO_2$; mol wt 339.48. C 77.83%, H 8.61%, N 4.13%, O 9.43%. Prepn of racemate: Pohland, Sullivan, *J. Am. Chem. Soc.* **75**, 4458 (1953); Pohland, U.S. pat. 2,728,779 (1955 to Lilly). Prepn of (+)-form: Pohland, Sullivan, *J. Am. Chem. Soc.* **77**, 3400 (1955). Stereochemistry: Sullivan *et al.*, *J. Org. Chem.* **28**, 2381 (1963); Casey, Myers, *J. Pharm. Pharmacol.* **16**, 455 (1964). Stereospecific synthesis: Pohland *et al.*, *J. Org. Chem.* **28**, 2483 (1963). Metabolism: S. L. Due *et al.*, *Biomed. Mass Spectrom.* **3**, 217 (1976). The α -dl- and d-diastereoisomers possess marked analgesic activity in contrast to the β -diastereoisomers which are substantially inactive. Toxicity: E. L. Goldenthal, *Toxicol. Appl. Pharmacol.* **18**, 185 (1971); J. I. Emerson *et al.*, *ibid.* **19**, 445 (1971). Comprehensive description: B. McEwan in *Analytical Profiles of Drug Substances* vol. 1, K. Florey, Ed. (Academic Press, New York, 1972) pp 301-318. Symposium on pharmacology, toxicology, and clinical efficacy of propoxyphene alone and in combination with acetaminophen: *Human. Toxicol.* **3**, Suppl., 1S-238S (1984).



Crystals from petr ether, mp 75-76°. $[\alpha]_D^{25} +67.3^\circ$ (c = 0.6 in chloroform).
 α -D-Hydrochloride, $C_{22}H_{30}ClNO_2$. *Algafan, Antalvic, Darvon, Depromic, Deprancol, Develin, Dolene, Dolocap, Doraphen, Erantin, Femadol, Harmar, Propox, Propoxychel, Proxagise*. Bitter crystals from methanol + ethyl acetate, mp 163-168.5°. $[\alpha]_D^{25} +59.8^\circ$ (c = 0.6 in water). Sol in water, alc, chloroform, acetone. Practically insol in benzene, ether. LD₅₀ in mice, rats (mg/kg): 28, 15 i.v.; 111, 58 i.p.; 211, 134 s.c.; 282, 230 orally (Emerson).
 α -D-Form napsylate monohydrate, $C_{32}H_{37}NO_5 \cdot S.H_2O$. *Darvon-N, Doloxene*. LD₅₀ orally in female rats: 990 mg/kg (Goldenthal).

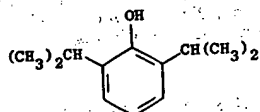
α-*l*-Form, see Levopropoxyphene.



mp 122°
chloride, $C_{17}H_{19}ClN_4O$, UP 106, Depressin, Vagran.

mp 235° CAT: Antidepressant.

1964). Propofol, 2,6-Bis(1-methylethyl)phenol; 2,6-diisopropylphenol; disoprofol; ICI 35868; Diprivan; Disoprivan; Rapinovet. $C_{12}H_{18}O$; mol wt 178.27. C 80.85%, H 10.18%, O 8.97%. Prepn: A. J. Kolka et al., *J. Org. Chem.* 21, 712 (1956); 22, 642 (1957); G. G. Ecke, A. J. Kolka, U.S. Pat. 2,831,898 (1958 to Ethyl Corp.); T. J. Kealy, D. D. Coffman, *J. Org. Chem.* 26, 987 (1961); B. E. Firth, T. J. Rosen, U.S. Pat. 4,447,657 (1984 to Universal Oil Products). Chromatographic study: J. K. Carlton, W. C. Bradbury, *J. Pharm. Chem. Soc.* 78, 1069 (1956). Animal studies: J. B. Glen, *Brit. J. Anaesth.* 52, 731 (1980). Pharmacokinetics: H. K. Adam et al., *ibid.* 743; *idem.* *ibid.* 55, 97 (1983). Determination in blood: *idem.*, *J. Chromatog.* 223, 232 (1981). Comparative studies vs other injectable anesthetics: B. Kay, *Comparative studies with other injectable anesthetics: B. Kay, D. K. Stephenson, Anaesthesia* 35, 1182 (1980); D. V. Rutter, *ibid.* 1188. Use in i.v. anesthesia: E. Major et al., *ibid.* 37, 541 (1982). Cardiovascular effects: D. Al-Khudhairi et al., *ibid.* 14007. Pharmacology of emulsion formulation: J. B. Glen, S. C. Hunter, *Brit. J. Anaesth.* 56, 617 (1984). Series of articles on pharmacology and clinical experience: *Postgrad. Med. J.* 61, Suppl. 3, 1-169 (1985).

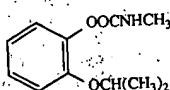


bp₃₀ 136°. bp₁₇ 126°. mp 19°. n_D²⁰ 1.5134. n_D²⁵ 1.5111. d₂₀ 0.955.

0.955.
THERAP CAT: Anesthetic (intravenous).
THERAP CAT (VET): Intravenous anesthetic (dogs and cats).

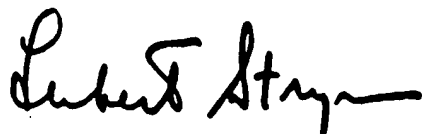
7848. Propolis. Bee bread; hive dross. A resinous substance found in beehives. Collected by bees from buds. Isola of caffeic acid from propolis: Cizmárik, Matel, *Experientia* 26, 713 (1970). Antimicrobial constituents of propolis: J. Metzner et al., *Pharmazie* 30, 799 (1975); E. M. Schneiderwind et al., *ibid.* 803. Review on the origin, chemical constituents and therapeutic activity: M. H. Haydak, *State of the Art Repts. State Apiarist* 1953, p. 74-87; M. Vanhaelen, R. Vanhaelen-Fastre, *J. Pharm. Belg* 34, 253 (1979). Greenish-brown, sticky mass. Aromatic odor. d 1.2. mp 64°. Becomes brittle when cooled below 15°. Extraction with alcohol gives *propolis wax*. The residue from the alcohol extraction is called *propolis resin*, yielding *propolis balsam* on extraction with hot petr ether. Propolis balsam has a hyacinth odor and is said to contain 10% cinnamyl alcohol.

849. Propoxur. 2-(1-Methylethoxy)phenol methylcarbamate; *o*-isopropoxyphenyl N-methylcarbamate; apocarb; BAY 39007; BAY 9010; Baygon; Bifex; Blattanex; Invisi-Gard; Propyon; Suncide; Sandran; Uden. $C_{11}H_{15}NO_3$; mol wt 209.24. C 63.14%, H 7.23%, N 6.69%, O 22.94%. Prep'n: U.S. pat. 3,111,539 (1963 to Bayer; Chemagro Corp.). Properties: Pflanzenschutz Nachr. Bayer 18, 53 (1965). Toxicity data: T. B. Gaines, Toxicol. Appl. Pharmacol. 14, 615 (1969). Teratogenicity study: K. D. Courtney et al., J. Environ. Sci. Health B20, 373 (1985).



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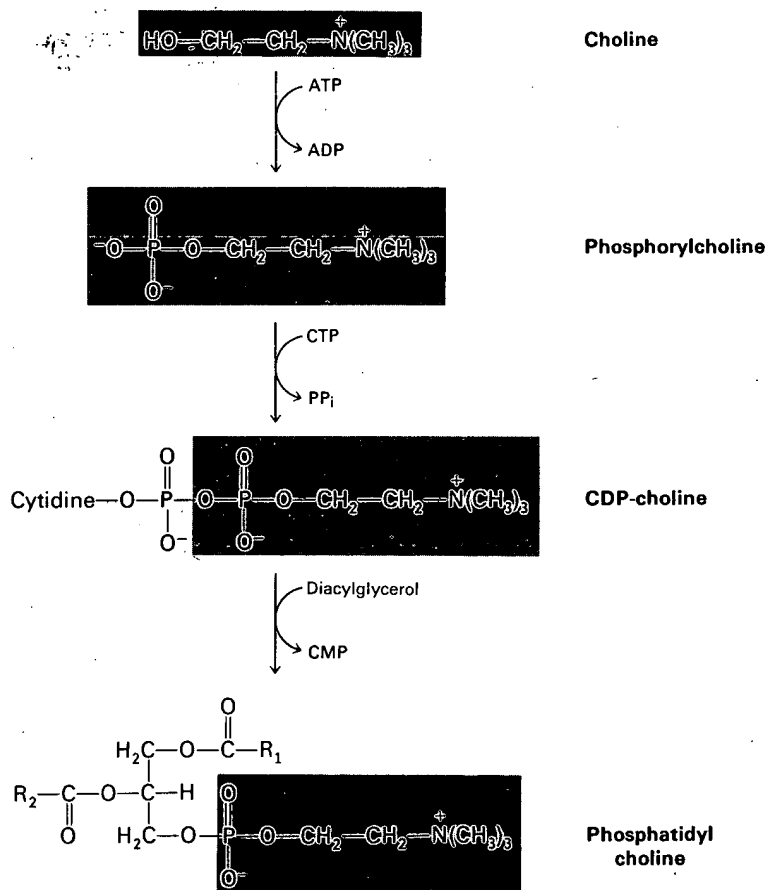
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**PHOSPHOGLYCERIDES CAN ALSO BE SYNTHESIZED
FROM A CDP-ALCOHOL INTERMEDIATE**

In mammals, phosphatidyl choline is synthesized by a pathway that utilizes choline obtained from the diet (Figure 23-2). Choline is phosphorylated by ATP to *phosphorylcholine*, which then reacts with CTP to form *CDP-choline*. The phosphorylcholine unit of CDP-choline is then transferred to a diacylglycerol to form *phosphatidyl choline*. Note that the activated species in this pathway is the cytidine derivative of phosphorylcholine rather than of phosphatidate.

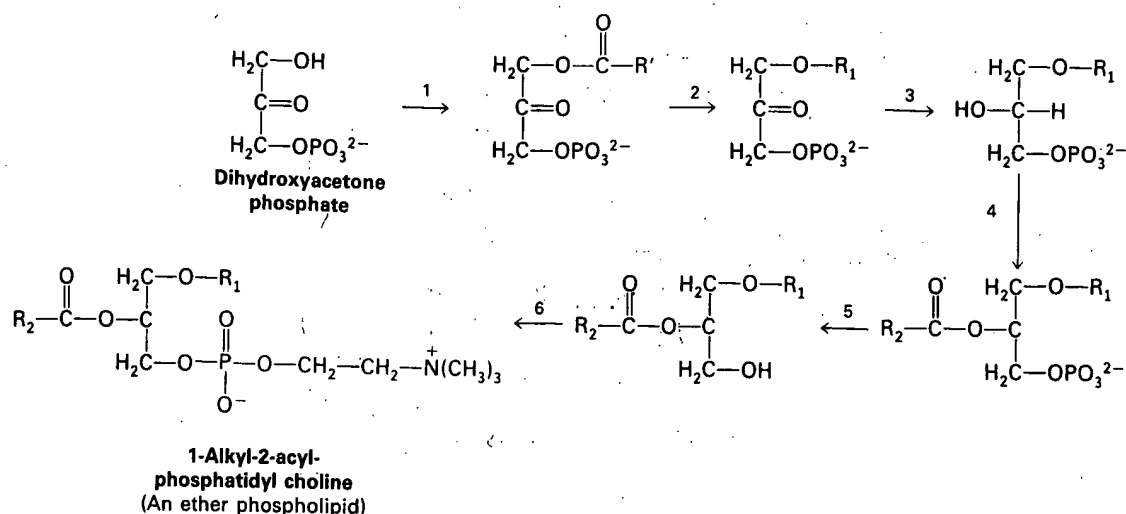
Figure 23-2
Synthesis of phosphatidyl choline.



Likewise, *phosphatidyl ethanolamine* can be synthesized from ethanolamine by forming a CDP-ethanolamine intermediate by analogous reactions. Alternatively, phosphatidyl ethanolamine can be formed from phosphatidyl serine by the enzyme-catalyzed exchange of ethanolamine for the serine moiety of the phospholipid.

**PLASMALOGENS AND OTHER ETHER PHOSPHOLIPIDS
ARE FORMED FROM DIHYDROXYACETONE PHOSPHATE**

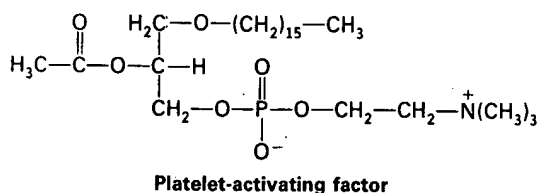
Some phospholipids contain an ether unit instead of an acyl unit at C_1 . *Glycerol ether phospholipids* are synthesized starting with dihydroxyacetone phosphate (Figure 23-3). Acylation by a fatty acyl CoA yields a 1-acyl derivative that exchanges with a long-chain alcohol to form an ether at C_1 . The keto group at C_2 is reduced by NADPH, and the

**Figure 23-3**

Synthesis of an ether phospholipid. The steps are (1) acylation by fatty acyl CoA, (2) exchange of an alcohol for the carboxylate moiety, (3) reduction by NADPH, (4) acylation by a second fatty acyl CoA, (5) hydrolysis of the phosphate ester, (6) transfer of a phosphocholine moiety.

resulting alcohol is acylated by a long-chain CoA. Removal of the 3-phosphate group yields 1-alkyl-2-acylglycerol, which reacts with CDP-choline to form the ether analog of phosphatidyl choline.

An ether phospholipid with striking activities has recently been identified. *Platelet-activating factor* is a 1-alkyl-2-acetyl ether analog of phosphatidyl choline. Even a very low concentration of this compound (0.1 nM) in the blood causes the aggregation of platelets and the dilation of blood vessels. The presence of an acetyl group rather than a long-chain acyl group at C₂ increases the water-solubility of this lipid, enabling it to function in an aqueous environment.



Plasmalogens are phospholipids containing an α,β -unsaturated ether at C₁. Phosphatidal choline, the plasmalogen corresponding to phosphatidyl choline, is formed by desaturation of a 1-alkyl precursor. The desaturase catalyzing this final step in the synthesis of a plasmalogen is a microsomal enzyme akin to the one that introduces double bonds into long-chain fatty acyl CoAs: O₂ and NADH are reactants, and cytochrome *b*₅ participates in catalysis (p. 489).

